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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/651,031	08/29/2003	Kazuo Ohkouchi	2003_1222A	8535
513	7590 04/18/2005		EXAMINER	
	TH, LIND & PONAC	SASTRI, SATYA B		
2033 K STRE SUITE 800	EET N. W.	ART UNIT	PAPER NUMBER	
WASHINGTON, DC 20006-1021			1713	

DATE MAILED: 04/18/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

S. Palerii and Trademark Office PTOL-326 (Rev. 1-04)	Office Action	Cummanı	Part of Paper No./Mail Date				
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing F 3) Information Disclosure Statement(s) (PTO Paper No(s)/Mail Date 1/23/04. S. Patent and Trademark Office		Paper N	v Summary (PTO-413) o(s)/Mail Date f Informal Patent Application (PTO-1 	152)			
 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 							
2. Certified copies of the priority documents have been received in Application No							
1. Certified copies of the priority documents have been received.							
a)⊠ All b)□ Some * c)□ No							
12)⊠ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).							
Priority under 35 U.S.C. § 119							
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
10)⊠ The drawing(s) filed on <u>29 August 2003</u> is/are: a)⊠ accepted or b)□ objected to by the Examiner.							
9) The specification is objected							
Application Papers							
	o resulction and/or el	ecaon requirement.					
	7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or election requirement.						
6) Claim(s) <u>1-6</u> is/are rejected.							
5) Claim(s) is/are allowed.							
4a) Of the above claim(s) is/are withdrawn from consideration.							
4)⊠ Claim(s) <u>1-6</u> is/are pending i	n the application.						
Disposition of Claims							
closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.							
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is							
2a)☐ This action is FINAL . 2b)☒ This action is non-final.							
1) Responsive to communication							
Status							
THE MAILING DATE OF THIS CO - Extensions of time may be available under the after SIX (6) MONTHS from the mailing date o - If the period for reply specified above is less th - If NO period for reply is specified above, the m - Failure to reply within the set or extended perion any reply received by the Office later than three earned patent term adjustment. See 37 CFR	MMUNICATION. provisions of 37 CFR 1.136(a) f this communication. Ian thirty (30) days, a reply with laximum statutory period will aj od for reply will, by statute, cau e months after the mailing date). In no event, however, may nin the statutory minimum of t pply and will expire SIX (6) M se the application to become	a reply be timely filed hirty (30) days will be considered timely. ONTHS from the mailing date of this com ABANDONED (35 U.S.C. § 133).	nmunication.			
A SHORTENED STATUTORY PE	RIOD FOR REPLY IS	S SET TO EXPIRE 3	MONTH(S) FROM				
The MAILING DATE of this of Period for Reply	communication appear	s on the cover sheet	with the correspondence add	ress			
		atya B. Sastri	1713				
Office Action Summ	1204	xaminer	Art Unit				
	1	0/651,031	OHKOUCHI ET AL.				
	A	pplication No.	Applicant(s)				

DETAILED ACTION

1. This office action is in response to application filed on August 29, 2003. *Claims 1-6* are now pending in the application.

Claim Objection

2. Claim 1 is objected to because of the following informalities: the claim language is vague, as the method does not recite the various steps involved in a clear and concise manner. Appropriate correction is required.

Claim Rejections - 35 USC § 103

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 4. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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5. Claims 1-6 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sakamoto et al. (US 5,785,821) in view of Mettetal et al. (US 4,358,347).

Sakamoto et al. disclose purification of acrylic acid from acrylic acid solution obtained after catalytic gas phase oxidation of propylene and/or acrolein. The aqueous acrylic acid is subjected to azeotropic distillation in the presence of poor water-soluble solvent (abstract).

Examples of such solvents disclosed include aliphatic and aromatic hydrocarbons having 7-8 carbon atoms (column 4, lines 14-37). The aqueous acrylic acid solution generally contains 50-80% by weight of acrylic acid, 1-5% by weight of acetic acid and 20-40% by weight of water under the ordinary conditions of acrylic acid synthesis. The proportions of these components may vary depending upon the operating conditions of the oxidation reactor and/or acrylic acid collecting tower (column 3, lines 19-27). The temperature of the top of the acrylic acid-collecting tower is 50-70°C (column 3, lines 46-48). The solution drawn out of the bottom of the azeotropic distillation column contains predominantly acrylic acid with trace amounts of acetic acid and other substances (Example 2).

The difference between the prior art and the instant invention is that the prior art does not teach the removal of glyoxal present as impurity in the aqueous solution along with acrylic acid from the distillation tower.

Evidence that carbonyl compounds are known by-product impurities during production of unsaturated acids from olefins and molecular oxygen is provided by the secondary reference (column 2, lines 25-29). Since the primary reference teaches the process of removal of acrylic acid by azeotropic distillation, it would have been obvious to one of ordinary skill in the art at

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the time the invention was made to remove impurity such as glyoxal present in acrylic acid and thereby obtain the present invention.

6. Claims 1-4, 6 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sakamoto et al. (EP 0861820 A2) as evidence provided by Mettetal et al. (US 4,358,347).

The prior art to Sakamoto et al. discloses a method of recovering acrylic acid produced by gas phase catalytic oxidation of propylene and/or acrolein. The acrylic acid is collected in the form of an aqueous solution and introduced into an azeotropic separation column and distilling in the presence of an azeotropic solvent to isolate and recover the acrylic acid. The azeotropic solvent may be either a mixed solvent composed of solvent A (ethyl acrylate, methyl methacrylate, isopropenyl acetate etc) and solvent B (toluene, heptane, 1-heptene, cycloheptene etc.) or the solvent A alone (abstract). The disclosed temperature at the top of the column ranges from 45-55°C and temperature at the bottom of the column ranged from 100-110°C (column 7, lines 1-8).

The difference between the prior art and the instant invention is that the prior art does not teach the removal of glyoxal present as impurity in the aqueous solution along with acrylic acid from the distillation tower.

Evidence that carbonyl compounds are known by-product impurities during production of unsaturated acids from olefins and molecular oxygen is provided by the secondary reference (column 2, lines 25-29). Since the primary reference teaches the process of removal of acrylic acid by azeotropic distillation, it would have been obvious to one of ordinary skill in the art at

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the time the invention was made to remove impurity such as glyoxal present in acrylic acid and

thereby obtain the present invention.

Conclusion

7. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Satya Sastri at (571) 212 1112.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, David Wu can be reached at (571) 212 1114.

The fax phone number for the organization where this application or proceeding is

assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

may be obtained from either Private PAIR or Public PAIR. Status information for unpublished

applications is available through Private PAIR only. For more information about the PAIR

system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private

PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

SATYA SASTRI

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April 1, 2005

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